C6—C11	1.379 (9)	C19-C20	1.52(1)
C7C8	1.371 (8)	C20-C21	1.52(1)
C8—C9	1.378 (9)	C21-C22	1.41(1)
C9-C10	1.382 (8)	C22-C23	1.55(1)
C9C12	1.447 (8)		
C5—O2—C6	116.9 (7)	N1-C13-C18	132.7 (8)
C12—O3—C14	104.5 (6)	C14-C13-C18	117.8 (9)
C16—O4—C19	120.8 (7)	O3-C14-C13	107.1 (9)
C12—N1—C13	105.2 (7)	O3-C14-C15	126.8 (8)
01–C5–O2	123.5 (9)	C13-C14-C15	126.1 (9)
01–C5–C4	126.2 (9)	C14-C15-C16	113.5 (7)
O2-C5-C4	110.3 (8)	O4—C16—C15	116.0 (7)
O2—C6—C7	120.7 (9)	O4-C16-C17	120.5 (9)
O2-C6-C11	117.8 (9)	C15-C16-C17	123.4 (9)
C7–C6–C11	121.4 (8)	C16-C17-C18	120.5 (8)
C6—C7—C8	118.6 (9)	C13C18C17	118.7 (7)
C7—C8—C9	120.2 (8)	O4—C19—O5	122(1)
C8—C9—C10	120.7 (8)	O4-C19-C20	108.9 (9)
C8-C9-C12	119.9 (9)	O5-C19-C20	129(1)
C10-C9-C12	119.3 (9)	C1C2C3	115. (1)
C9C10C11	119.6 (8)	C2-C3-C4	111.4 (9)
C6-C11-C10	119.4 (9)	C3-C4-C5	113.1 (8)
O3-C12-N1	113.8 (8)	C19-C20-C21	116.5 (9)
O3-C12-C9	117.9 (8)	C20C21C22	111. (1)
N1-C12-C9	128.2 (9)	C21-C22-C23	109. (1)
N1-C13-C14	109.4 (8)		
C6-02-C5-01	2.6 (9)	O2-C6-C11-C10	-179.2 (5
C6-02-C5-C4	-177.4 (5)	C7-C8-C9-C12	174.7 (5
C5-02-C6-C7	70.9 (7)	C12-C9-C10-C11	-174.2 (5
C5-02-C6-C11	-110.7 (6)	C8-C9-C12-O3	1.7 (8
C19-04-C16-C15	107.6 (6)	C8-C9-C12-N1	-173.4 (6
C19—O4—C16—C17	-76.6 (7)	C10C9C12O3	177.7 (5
C16—O4—C19—O5	-1.8 (9)	C10-C9-C12-N1	2.6 (9
C16-04-C19-C20	-179.3 (5)	N1-C13-C18-C17	177.5 (6
C13-N1-C12-C9	174.4 (5)	C14-C15-C16-O4	171.8 (5
C1-C2-C3-C4	177.7 (7)	O4-C16-C17-C18	-171.5 (5
C2-C3-C4-C5	174.1 (6)	O4-C19-C20-C21	-161.6 (5
C3-C4-C5-01	-7(1)	O5-C19-C20-C21	21 (1)
C3-C4-C5-O2	173.5 (6)	C19-C20-C21-C22	72.9 (9
02	179.6 (5)	C20-C21-C22-C23	175.4 (8

H atoms were placed at calculated positions [riding model, C—H = 0.95 Å, $U(H) = U_{eq}(C)$]. A refinement against 3037 unique reflections having I > 0 gave R = 0.211, wR = 0.091and S = 1.305 (weights as above), without significant changes in the final structural parameters. Program used to solve structure: *MULTAN11*/82 (Main, Fiske, Hull, Lessinger, Germain, Declercq & Woolfson, 1982). Programs used to refine structure: *SDP* (Enraf-Nonius, 1985). Refinement was by full-matrix least-squares method. All programs were run on a Digital MicroVAX computer.

The author thanks Professor Augusto Sirigu for helpful discussions. This work was supported by the Ministero dell' Università e della Ricerca Scientifica e Tecnologica.

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Acta Cryst. (1994). C50, 263-265

1,3,5-Triethyl-1,3,5-triazine-2,4,6(1*H*,3*H*,5*H*)-trione, C₉H₁₅N₃O₃

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(Received 17 May 1993; accepted 13 August 1993)

Abstract

The crystal structure of the title compound (triethylisocyanuric acid), which was obtained from the preparation of an ethyl isocyanate derivative of the anti-tumor compound CC-1065, has been determined by X-ray diffraction.

Comment

The title compound (I) was prepared accidentally during the synthesis of a derivative of the known anti-tumor com-



pound CC-1065 (Chidester, Krüger, Mizsak, Duchamp & Martin, 1981). The attempted synthesis of an ethyl isocyanate derivative of CC-1065 yielded the crystalline compound triethylisocyanuric acid. The crystals appeared as clear yellow prisms from a reaction mixture of ethyl isocyanate and pyridine. The perhydrotriazine moiety is essentially planar with the terminal methyl groups lying

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 71446 (29 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: NA1045]

1.4 Å out of the plane. The molecular packing in the unit cell viewed down the b axis is shown in Fig. 2. The bond angles and distances are in close agreement with those of trimethylisocyanuric acid (Belaj & Nachbaur, 1987).



Fig. 1. View of $C_9H_{15}N_3O_3$ showing the labeling of the atoms.



Fig. 2. Packing drawing of the title molecule along the b axis.

Experimental

Crystal data C₉H₁₅N₃O₃ $M_r = 213.24$ Orthorhombic $P2_12_12_1$ a = 7.845 (1) Å b = 8.160(1) Å c = 16.840(1) Å

 $V = 1078.0 (1) \text{ Å}^3$

Cu $K\alpha$ radiation
$\lambda = 1.5418 \text{ Å}$
Cell parameters from 25
reflections
$\theta = 60 - 69^{\circ}$
$\mu = 0.748 \text{ mm}^{-1}$
T = 123 K
Thin plate

tion in CsCl₂/H₂O

Data collection

Siemens P1 diffractometer $\theta/2\theta$ scans Absorption correction: none 1183 measured reflections 1183 independent reflections 942 observed reflections $[I > 3.0\sigma(I)]$

Refinement

C(2)

Refinement on F^2 R(F) = 0.035 $wR(F^2) = 0.078$ S = 1.571183 reflections 195 parameters H-atom parameters not refined $w = 1/\sigma^2(F^2)$

$$\begin{array}{l} 0.50 \times \ 0.28 \times \ 0.16 \ \text{mm} \\ \text{Clear} \end{array}$$

 $\theta_{\rm max} = 69^{\circ}$ $h = 0 \rightarrow 9$ $k = 0 \rightarrow 9$ $l = 0 \rightarrow 20$ 9 standard reflections monitored every 80 reflections intensity variation: none

 $(\Delta/\sigma)_{\rm max} = 0.027$ $\begin{array}{l} \Delta\rho_{\rm max} = 0.172 \ {\rm e} \ {\rm \AA}^{-3} \\ \Delta\rho_{\rm min} = -0.220 \ {\rm e} \ {\rm \AA}^{-3} \end{array}$ Atomic scattering factors from Doyle & Turner (1968) (C, N, O) and Stewart, Davidson & Simpson (1965) (H)

Cell refinement: Duchamp (1977). Data reduction: CRYM (Duchamp, 1984). Program(s) used to solve structure: MUL-TAN (Main et al., 1980). Program(s) used to refine structure: CRYM.

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters ($Å^2$)

 $B_{\rm eq} = (8\pi^2/3) \sum_i \sum_j U_{ij} a_i^* a_i^* \mathbf{a}_i \cdot \mathbf{a}_j.$

x	у	z	Bea
0.1067 (3)	0.1951 (3)	0.6116 (1)	1.94 (9)
0.0574 (5)	0.0214 (4)	0.6212 (2)	2.7 (1)
0.3263 (3)	0.1599 (3)	0.6971 (1)	2.90 (9)
0.2463 (4)	0.2494 (3)	0.6538(1)	2.2 (1)
0.2890 (3)	0.4127 (3)	0.6441 (1)	1.94 (9)
0.4403 (4)	0.4758 (4)	0.6857 (2)	2.5(1)
0.2457 (3)	0.6602 (2)	0.5855(1)	2.94 (9)
0.2024 (4)	0.5188 (4)	0.5940 (1)	2.2 (1)
0.0625 (3)	0.4538 (3)	0.5551 (1)	2.05 (9)
-0.0273 (4)	0.5627 (4)	0.4988 (2)	2.4 (1)
-0.1066 (3)	0.2375 (2)	0.5213 (1)	2.55 (9)
0.0114 (4)	0.2915 (3)	0.5595 (1)	2.1 (1)
-0.0487 (5)	0.0049 (5)	0.6958 (2)	3.3 (1)
0.5998 (5)	0.4589 (5)	0.6353 (2)	3.4 (2)
0.0677 (5)	0.5766 (4)	0.4206 (2)	2.9(1)

Table 2. Selected geometric parameters (Å, °)

	•	-	• • •
N(1)—C(1)	1.478 (4)	C(3)-C(33)	1.518 (5)
N(1)-C(2)	1.378 (4)	O(4)-C(4)	1.211 (4)
N(1)C(6)	1.396 (4)	C(4)N(5)	1.383 (4)
C(1) - C(11)	1.523 (5)	N(5)—C(5)	1.478 (4)
O(2) - C(2)	1.208 (3)	N(5)-C(6)	1.386 (4)
C(2)—N(3)	1.384 (4)	C(5)-C(55)	1.518 (4)
N(3)-C(3)	1.471 (4)	O(6)—C(6)	1.210 (3)
N(3)C(4)	1.386 (4)		
C(1) - N(1) - C(2)	117.4 (2)	N(3)-C(4)-O(4)	121.9 (3)
C(1) - N(1) - C(6)	118.0 (2)	N(3) - C(4) - N(5)	116.0 (2)
C(2) - N(1) - C(6)	124.6 (2)	O(4)-C(4)-N(5)	122.1 (3)
N(1) - C(1) - C(11)	111.7 (3)	C(4) - N(5) - C(5)	116.8 (2)
N(1) - C(2) - O(2)	122.0 (3)	C(4) - N(5) - C(6)	124.8 (2)
N(1)-C(2)-N(3)	116.2 (2)	C(5)-N(5)-C(6)	118.1 (2)

O(2) - C(2) - N(3)	121.9 (3)	N(5)-C(5)-C(55)	111.5 (3)
C(2) - N(3) - C(3)	118.4 (2)	N(1)-C(6)-N(5)	114.6 (2)
C(2) - N(3) - C(4)	123.7 (2)	N(1)-C(6)-O(6)	122.6 (3)
C(3) - N(3) - C(4)	117.8 (2)	N(5)-C(6)-O(6)	122.8 (3)
N(3)-C(3)-C(33)	111.5 (2)		

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 71553 (11 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: CD 1063]

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Acta Cryst. (1994). C50, 265-267

Crystal Studies of Musk Compounds. V.† Structure of 1-(1,2,6,7,8,8a-Hexahydro-3,6,6,8a-tetramethylacenaphthylen-4-yl)ethan-1-one

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(Received 4 May 1993; accepted 30 June 1993)

Abstract

The acetyl group in the title compound, $C_{18}H_{24}O$, is twisted by 31.6 (1)° out of the plane of the aromatic ring with its O atom facing the neighbouring methyl group. The aromatic ring is essentially planar; due to the strain induced by the fused rings, the non-aromatic six- and fivemembered rings adopt deformed half-chair and envelope conformations, respectively.

Comment

This paper reports the only tricyclic structure determined in this series dealing with crystal studies of musk compounds (Part IV: De Ridder & Schenk, 1992).

Recently, Fehr, Galindo, Haubrichs & Perret (1989) used the results of the structure-activity relationship studies of Beets (1978) and those of Theimer & Davies (1967) to construct new strong musks. Fehr and co-workers arrived at the basic musk skeleton given in Fig. 1.



Fig. 1. Basic musk skeleton structure, R = H, CH₃ (reprinted from Fehr *et al.*, 1989).

This skeleton is represented in a musk compound described in the second paper of this series (De Ridder, Goubitz & Schenk, 1990): incorporation of a methyl group at the position marked 'c' leads to musk tonalid ($R = CH_3$). By the appropriate structural modification of known musk odorants, Fehr, Galindo, Haubrichs & Perret (1989) manufactured a number of new compounds. According to these authors, the incorporation of an additional CH₃ or CH₂ group into the basic musk skeleton of Fig. 1 modifies only slightly the global shape of the molecule, but leads to densely packed structures of enhanced lipophilicity. *Via* the incorporation of a methyl group at the position marked 'a', a tricyclic musk (I) was obtained, the crystal structure of which is described in this paper.



The title compound is a strong musk and was obtained from the Firmenich laboratories.

For convenience, the six-membered rings are numbered analogously to the tetralin structures described by De Ridder, Goubitz & Schenk (1990) (see Fig. 2). This numbering is different to the commonly used IUPAC system.

The phenyl ring is practically planar; the atoms directly attached to the benzene ring show considerable devia-

[†] This work forms part of a thesis by De Ridder (1992).

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